

- [27] *N. Yanaihara, T. Hashimoto, C. Yanaihara, K. Tsuji, Y. Kenmochi, F. Ashizawa, T. Kaneko, H. Oka, S. Saito, A. Arimura & A. V. Schally*, *Biochem. biophys. Res. Commun.* **52**, 64 (1973).
- [28] *D. H. Coy, E. J. Coy & A. V. Schally*, *J. med. Chemistry* **16**, 83 (1973).
- [29] *M. Monahan, J. Rivier, W. Vale, N. Ling, G. Grant, M. Amoss, R. Guillemin, R. Burgus, E. Nicolaides & M. Rebstock*, 'Chemistry and Biology of Peptides', J. Meienhofer ed., p. 601, Ann Arbor, Science Publishers, Ann Arbor, 1972.
- [30] *J. Chang, R. H. Williams, A. J. Humphries, N. G. Johansson & K. Folkers*, *Biochem. biophys. Res. Commun.* **47**, 727 (1972).
- [31] *C. I. Bliss*, 'The Statistics of Bioassay', Academic Press Inc., New York, 1952.
- [32] *B. Kerdelhué, A. Bérault, C. Courte & M. Jutisz*, *C. R. hebdom. Seances Acad. Sci. (Paris) Ser. D.* **269**, 2413 (1969).
- [33] *B. Kerdelhué, G. Kann & M. Jutisz*, 'Hormones Glycoprotéiques' Colloque INSERM, M. Jutisz ed., p. 177, published by INSERM, Paris, 1972.
- [34] *B. Kerdelhué & M. Jutisz*, *Abstr. IVth Int. Congr. Endocrinol. Washington D.C., Excerpta Medica I.C.S.* **256**, 141 (1972), (Abstr. No. 352).
- [35] *D. H. Spackman, W. H. Stein & S. Moore*, *Analyt. Chemistry* **30**, 1190 (1958).
- [36] *T. Y. Lin & Y. H. Chang*, *J. biol. Chemistry* **246**, 2842 (1971).
- [37] IUPAC-IUB Commission, *Biochemistry* **5**, 2485 (1966); *ibid.* **6**, 362 (1967).
- [38] *P. Karrer, M. Gisler, E. Horlacher, F. Locher, W. Mäder & H. Thomann*, *Helv.* **5**, 482 (1922).
- [39] *W. Siedel, K. Sturm & R. Geiger*, *Chem. Ber.* **96**, 1436 (1963).
- [40] *E. Kliegler & H. Gibian*, *Liebigs Ann. Chem.* **649**, 183 (1961).
- [41] *W. R. Gray*, 'Methods in Enzymology', C.H.W. Hirs ed., Vol. XI, 139, Academic Press, New York, 1967.
- [42] *E. Waser & E. Brauchli*, *Helv.* **7**, 740 (1924).
- [43] *M. Bergmann & L. Zervas*, *Chem. Ber.* **65**, 1192 (1932).

233. Dicationic Diolefinic Complexes of Palladium(II) and Platinum(II)

Preliminary communication

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Uncharged diolefinic complexes $M(\text{diolefin})\text{Cl}_2$ (**1**; $M = \text{Pt(II)}, \text{Pd(II)}$) have been obtained by *Chatt et al.* in 1957 [1]. The coordinated diolefins are susceptible of nucleophilic attack by alcohols, amines, etc. [1–4]. In view of obtaining complexes in which the double bond is prone to be attacked by weaker nucleophiles and to give thermally 'forbidden' [2+2]-cycloadditions, we have prepared the dicationic complexes $[M(\text{diolefin})(\text{CH}_3\text{CN})_2](\text{PF}_6)_2$ (**2**; diolefin = cycloocta-1,5-diene, 2,5-norbornadiene) and $[\text{Pd}(\text{cod})\text{L}](\text{PF}_6)_2$ (**3**; $\text{L} = 2,2'$ -dipyridyl, bis(1,2-diphenylphosphine)ethane).

The white complexes **2** are obtained in the solid state by adding a stoichiometric amount of $[\text{Ag}(\text{CH}_3\text{CN})_2]\text{PF}_6$ to a dichloromethane solution of **1** at -15° , extracting **2** from AgCl with acetone and adding acetonitrile then dichloromethane. These complexes are stable when stored at -15° . IR.-spectra show that acetonitrile is coordinated ($\nu(\text{CN}) = 2328\text{--}2340\text{ cm}^{-1}$, compared to 2266 for free CH_3CN and 2347

for $\text{Pd}(\text{CH}_3\text{CN})_4^{2+}$ [5]); $\nu(\text{CC})$ and $\delta(\text{CH})$ are about the same as those of complexes **1**. $^1\text{H-NMR}$ -spectra show a deshielding of olefinic protons compared to those in the corresponding complexes **1**. Complexes **2** with dicyclopentadiene are more soluble and nucleophilic addition of acetonitrile on the coordinated nitrile occurred prior to isolation. IR.- and $^1\text{H-NMR}$ -spectra indicate that the corresponding dicationic imino compounds are formed ($\nu(\text{NH}) = 3360 \text{ cm}^{-1}$, $\nu(\text{C}=\text{N}) = 1638 \text{ cm}^{-1}$).

Complexes **2**, for example with cyclooctadiene, react with methanol or *t*-butylalcohol to give the 2-alkoxycyclooct-5-enyl derivatives. Complex **3** ($\text{M} = \text{Pd}$, $\text{L} = 2,2'$ -dipyridyl) reacts with methanol in the presence of sodium carbonate to give metallic Pd and 1- and 2-methoxycycloocta-1,3-dienes. Thus, as the four sites of coordination of **3** are occupied by ligands not displaced by CH_3OH (or CH_3O^-), we conclude that the nucleophilic attack occurs directly at the coordinated olefin.

Complexes **2**, having two weakly coordinated ligands in *cis*-positions to the diene which is activated by the high positive charge on the metal, could promote cycloadditions with olefins. Indeed $[\text{Pd}(\text{nbd})(\text{CH}_3\text{CN})_2](\text{PF}_6)_2$ in acetonitrile reacts with excess of propylene at -15° . The mass- and $^1\text{H-NMR}$ -spectra of the isolated organic product extracted with pentane show that $[2+2]$ - and $[2+4]$ -cycloadditions of propylene on norbornadiene occurred without dimerization of norbornadiene itself. The product of one thermally 'forbidden' $[2+2]$ -cycloaddition, 3-methyltricyclo[4.2.1.0^{2,5}]nonene-7, formed 55% of the extract. The remaining components were mainly the dimers ($\sim 30\%$) and trimers ($\sim 10\%$), 3,8 or 9-dimethyltetracyclo[4.4.1.0^{2,5}.0^{7,10}]undecane, 6,10 or 11-dimethyltetracyclo[6.3.0.0^{2,9}.0^{4,8}]undecane, trimethylpentacyclo[6.4.1.0^{2,7}.0^{3,6}.0^{9,12}]tridecane and trimethylpentacyclo[8.3.0.0^{2,11}.0^{4,10}.0^{6,9}]tridecane. We did not observe any reaction of complexes **2** with acetylene or acetylene-dimethylcarboxylate.

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REFERENCES

- [1] *J. Chatt, L. M. Vallarino & L. M. Venanzi*, *J. chem. Soc.* 1957, 3413.
- [2] *R. G. Schultz*, *J. Organometal. Chemistry* 6, 435 (1966).
- [3] *J. K. Stille & R. A. Morgan*, *J. Amer. chem. Soc.* 88, 5135 (1966).
- [4] *F. R. Hartley*, 'The Chemistry of Platinum and Palladium', Applied Science Publishers Ltd., London, 1973, p. 386.
- [5] *B. B. Wayland & R. F. Schramm*, *Inorg. Chemistry* 8, 971 (1969).